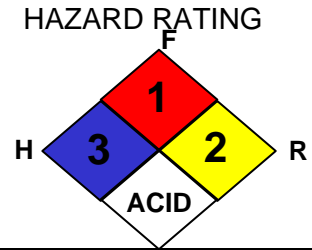




CONCORDE BATTERY LEAD ACID BATTERY



MATERIAL SAFETY DATA SHEET

SECTION 1 – CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

MANUFACTURER'S NAME: CONCORDE BATTERY CORPORATION	EMERGENCY TELEPHONE NO.: CHEMTEL 800-255-3924
ADDRESS: 2009 San Bernardino Rd., West Covina, CA 91790	OTHER INFORMATION CALLS: 626-813-1234
PERSON RESPONSIBLE FOR PREPARATION: Jorge Gonzalez	Revised Date: November 10, 2011

SECTION 2 - COMPOSITION/INFORMATION ON INGREDIENTS

C.A.S.	PRINCIPAL HAZARDOUS COMPONENT(S) (chemical & common name(s))	Hazard Category	% Weight	ACGIH TLV	OSHA PEL/TWA
7439-92-1	Lead/Lead Oxide/Lead Sulfate	Acute-Chronic	55-70%	0.05 mg/m ³	0.05 mg/m ³
7440-36-0	Antimony	Chronic	0-5%	0.5 mg/m ³	0.5 mg/m ³
7440-38-2	Arsenic (inorganic)	Acute-Chronic	<1%	0.01 mg/m ³	0.01 mg/m ³
7664-93-9	Sulfuric Acid (Battery Electrolyte)	Reactive-Oxidizer Acute -Chronic	20-35%	1.0 mg/m ³	1.0 mg/m ³
7440-70-2	Calcium	Reactive	<0.15%	Not Established	Not Established
7440-31-5	Tin	Chronic	<0.3%	2 mg/m ³	2 mg/m ³

Note: PEL's for individual states may differ from OSHA's PEL's. Check with local authorities for the applicable state PEL's.
 OSHA – Occupational Safety and Health Administration; ACGIH – American Conference of Governmental Industrial Hygienists; NIOSH – National Institute for Occupational Safety and Health.

COMMON NAME: (Used on label) Lead-acid battery
 (Trade Name & Synonyms) Lead-Acid Storage Battery; Electric Storage Battery Chemical Family: Toxic and Corrosive Material Mixture
 Chemical Name: Lead-Acid Storage Battery Formula: Lead and Acid (electrolyte)

SECTION 3 -- HAZARD IDENTIFICATION

Signs and Symptoms of Exposure	1. Acute Hazards	Do not open battery. Avoid contact with internal components. Internal components include lead and liquid electrolyte. Electrolyte - Electrolyte is corrosive and contact may cause skin irritation and chemical burns. Electrolyte causes severe irritation and burns of eyes, nose and throat. Ingestion can cause severe burns and vomiting. Lead - Direct skin or eye contact may cause local irritation. Inhalation or ingestion of lead dust or fumes may result in headache, nausea, vomiting, abdominal spasms, fatigue, sleep disturbances, weight loss, anemia and leg, arm and joint pain.			
2. Subchronic and Chronic Health Effects	Electrolyte - Repeated contact with sulfuric acid and battery electrolyte fluid may cause drying of the skin that may result in irritations, dermatitis, and skin burns. Repeated exposure to sulfuric acid mist may cause erosion of teeth, chronic eye irritation and / or chronic inflammation of the nose, throat, and lungs. Lead - Prolonged exposure may cause central nervous system damage, gastrointestinal disturbances, anemia, irritability, metallic taste, insomnia, wrist-drop, kidney dysfunction and reproductive system disturbances. Pregnant women should be protected from excessive exposure to prevent lead from crossing the placental barrier and causing infant neurological disorders. California Proposition 65 Warning: Battery posts, terminals, and related accessories contain lead and lead compounds, chemicals known to the State of California to cause cancer and reproductive harm. During charging, strong inorganic acid mists containing sulfuric acid are evolved, a chemical known to the State of California to cause cancer. Wash hands after handling.				
Medical Conditions Generally Aggravated by Exposure	If battery is broken or material is spilled, then persons with the following medical conditions must take precautions: pulmonary edema, bronchitis, emphysema, dental erosion and tracheobronchitis.				
Routes of Entry	Inhalation - YES Ingestion – YES	Eye Contact- YES Skin Contact - YES			
Chemical(s) Listed as Carcinogen or potential Carcinogen	Proposition 65 - YES	National Toxicology Program - YES	I.A.R.C. Monographs - YES	OSHA - NO	

SECTION 4 - FIRST AID MEASURES

Emergency and First Aid Procedures	Contact with internal components if battery is opened/broken.
1. Inhalation	Remove to fresh air and provide medical oxygen/CPR if needed. Obtain medical attention.
2. Eyes	Immediately flush with water for at least 15 minutes, hold eyelids open. Obtain medical attention.
3. Skin	Flush contacted area with large amounts of water for at least 15 minutes. Remove contaminated clothing and obtain medical attention if necessary.
4. Ingestion	Do not induce vomiting. If conscious drink large amounts of water/milk. Obtain medical attention. Never give anything by mouth to an unconscious person.

SECTION 5 - FIREFIGHTING MEASURES

Flash Point	Not Applicable	Flammable Limits in Air % by Volume (When Charging)	Hydrogen (H ₂)	Lower 4.1%	Upper 74.2%	Extinguisher ABC, Media	Class CO ₂ , Halon	Auto-Ignition Temperature	Polypropylene 675°F
Special Fire Fighting Procedures	Lead/acid batteries do not burn, or burn with difficulty. Do not use water on fires where molten metal is present. Extinguish fire with agent suitable for surrounding combustible materials. Cool exterior of battery if exposed to fire to prevent rupture. The acid mist and vapors generated by heat or fire are corrosive. Use NIOSH approved self-contained breathing apparatus (SCBA) and full protective equipment operated in positive-pressure mode.								
Unusual Fire and Explosion Hazards	Hydrogen gas and sulfuric acid vapors are generated upon overcharge and polypropylene case failure. Ventilate charging areas as per ACGIH <u>Industrial Ventilation: A Manual of Recommended Practice</u> . Hydrogen gas may be flammable or explosive when mixed with air, oxygen, and chlorine. Avoid open flames/sparks/other sources of ignition near battery. To avoid risk of fire or explosion, keep sparks or other sources of ignition away from batteries and do not allow metallic materials to simultaneously contact negative and positive terminals of cells and batteries. SULFURIC ACID REACTS VIOLENTLY WITH WATER/ORGANICS.								

SECTION 6 - ACCIDENTAL RELEASE MEASURES

Procedures for Cleanup:	Avoid contact with any spilled materials. Contain spill, isolate hazard area, and deny entry. Limit site access to emergency responders. Neutralize with sodium bicarbonate, soda ash, lime or other neutralizing agent. Place battery in suitable container for disposal. Dispose of contaminated materials in accordance with applicable local, state, and federal regulations. Sodium bicarbonate, soda ash, sand, lime or other neutralizing agent should be kept on-site for spill remediation.
Personal Precautions:	Acid resistant gloves, aprons, boots, and protective clothing. ANSI approved chemical splash goggles w/face shield recommended. Ventilate enclosed areas.
Environmental Precautions:	Lead and its compounds and sulfuric acid can pose a severe threat to the environment. Contamination of water, soil and air should be prevented.

SECTION 7 - HANDLING AND STORAGE

Precautions to be Taken in Handling and Storage	Store away from reactive materials, open flames and sources of ignition. Combustion or overcharging may create or liberate toxic and hazardous gases and liquids including hydrogen, sulfuric acid mist, sulfur dioxide, sulfur trioxide, stibine, arsine and sulfuric acid. Store batteries in cool, dry well-ventilated area. Do not short circuit battery terminals, or remove vent caps during storage or recharging. Protect battery from physical damage and adverse weather conditions. Avoid contact with internal components.
Other Precautions	GOOD PERSONAL HYGIENE AND WORK PRACTICES ARE MANDATORY. Refrain from eating, drinking or smoking in work areas. Thoroughly wash hands, face, neck and arms, before eating, drinking and smoking. Launder soiled clothing before reuse. Emptied batteries contain hazardous sulfuric acid residue.

SECTION 8 - EXPOSURE CONTROLS AND PERSONAL PROTECTION

Respiratory Protection (Specify Type)	None required under normal conditions. An acid-gas NIOSH approved respirator is required when the PEL is exceeded or employee experiences respiratory irritation. When exposure levels are unknown or when firefighting, wear a self-contained breathing apparatus with a full facepiece operated in positive pressure mode.				
Ventilation	Store and handle in a dry, well ventilated area.	Local Exhaust	When PEL is exceeded.	Mechanical (General)	Normal mechanical ventilation recommended for stationary applications and when charging.
Protective Gloves	Wear rubber or plastic acid resistant gloves with elbow length gauntlet when filling batteries.	Eye Protection	ANSI approved safety glasses with side shields/face shield recommended. Use ANSI approved chemical splash goggles w/ face shield when filling battery with electrolyte.		
Other Protective Clothing or Equipment	Ventilation shall be provided in areas where exposures are above the PEL or TLV specified by OSHA or other local, state and federal regulations. Acid-resistant rubber or plastic apron, boots, and protective clothing. Safety shower and eyewash.				

SECTION 9 - PHYSICAL AND CHEMICAL PROPERTIES

Boiling Point	Electrolyte Approx. 235° F	Vapor Pressure	Electrolyte 1 mm Hg @ 145.8° F	Specific Gravity	Electrolyte (2O = 1) 1.250-1.320 pH <2	Melting Point	Polypropylene >320°F
Percent Volatile By Volume (%)	Not Applicable		Vapor Density	Hydrogen (Air =1): 0.069 Electrolyte (Air = 1): 3.4 @ STP		Evaporation Rate	Not applicable
Solubility In Water	Electrolyte 100% soluble (electrolyte)		Reactivity in Water Electrolyte – Water Reactive (1)				
Appearance and Odor: terminals.	Battery: Polypropylene or hard rubber case, solid; may be contained within an outer casing of aluminum or steel. Case has metal terminals. Lead (internal): Gray, metallic, solid; brown/grey oxide Electrolyte: Liquid, colorless, oily fluid; nuisance odor when hot or charging battery.						

SECTION 10 - STABILITY AND REACTIVITY

Stability:	Stable	Conditions to Avoid	High temperatures-cases decompose at >320°F. Avoid overcharging, smoking, sparks near battery surface and rapid overcharge.
Incompatibility (Materials to Avoid)	Sparks, open flames, keep battery away from strong oxidizers.		
Hazardous Decomposition Products	An explosive hydrogen/oxygen mixture within the battery may occur during charging. Combustion can produce carbon dioxide (CO ₂) and carbon monoxide (CO). Molten metals produce fumes and/or vapor that may be toxic or respiratory irritants.		
Hazardous Polymerization	Has not been reported.		Do not overcharge.

SECTION 11 - TOXICOLOGICAL INFORMATION

GENERAL: The primary routes of exposure to lead are ingestion or inhalation of dust and fumes.

ACUTE:

INHALATION/INGESTION: Exposure to lead and its compounds may cause headache, nausea, vomiting, abdominal spasms, fatigue, sleep disturbances, weight loss, anemia, and pain in the legs, arms and joints. Kidney damage, as well as anemia, can occur from acute exposure.

CHRONIC:

INHALATION/INGESTION: Prolonged exposure to lead and its compounds may produce many of the symptoms of short-term exposure and may also cause central nervous system damage, gastrointestinal disturbances, anemia, and wrist drop. Symptoms of central nervous system damage include fatigue, headaches, tremors, hypertension, hallucination, convulsions and delirium. Kidney dysfunction and possible injury has also been associated with chronic lead poisoning. Chronic over-exposure to lead has been implicated as a causative agent for the impairment of male and female reproductive capacity, but there is at present, no substantiation of the implication. Pregnant

women should be protected from excessive exposure. Lead can cross the placental barrier and unborn children may suffer neurological damage or developmental problems due to excessive lead exposure in pregnant women.

SECTION 12 - ECOLOGICAL INFORMATION

In most surface water and groundwater, lead forms compounds with anions such as hydroxides, carbonates, sulfates, and phosphates, and precipitates out of the water column. Lead may occur as sorbed ions or surface coatings on sediment mineral particles or may be carried in colloidal particles in surface water. Most lead is strongly retained in soil, resulting in little mobility. Lead may be immobilized by ion exchange with hydrous oxides or clays or by chelation with humic or fulvic acids in the soil. Lead (when in the dissolved phase) is bioaccumulated by plants and animals, both aquatic and terrestrial.

SECTION 13 - DISPOSAL CONSIDERATIONS

Waste Disposal Methods	Lead-acid batteries are completely recyclable. Return whole scrap batteries to distributor, manufacturer or lead smelter for recycling. For information on returning batteries to Concorde Battery for recycling call 626-813-1234. For neutralized spills, place residue in acid-resistant containers with sorbent material, sand or earth and dispose of in accordance with local, state and federal regulations for acid and lead compounds. Contact local and/or state environmental officials regarding disposal information.
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SECTION 14 - TRANSPORT INFORMATION

U.S. DOT PROPER SHIPPING NAME: Batteries, wet, filled with acid
U.S. DOT HAZARD CLASS: 8
U.S. DOT ID NUMBER: UN 2794
U.S. DOT PACKING GROUP: III
U.S. DOT LABEL: CORROSIVE

IMO PROPER SHIPPING NAME: Batteries, wet, filled with acid
IMO U.N. CLASS: 8
IMO U.N. NUMBER: UN 2794
IMO LABEL: CORROSIVE
IMO VESSEL STOWAGE: A
Ems # - F-A, S-B

IATA PROPER SHIPPING NAME: Batteries, wet, filled with acid
IATA U.N. CLASS: 8
IATA U.N. NUMBER: UN 2794
IATA LABEL: CORROSIVE
ERG Code: 8L

SECTION 15 - REGULATORY INFORMATION

U.S. Hazardous Under Hazard Communication Standard:

Lead – YES
Sulfuric Acid – YES
Antimony – YES
Arsenic - YES

Ingredient Listed on TSCA Inventory:

YES

CERCLA Section 304 Hazardous Substances:

Lead– YES

RQ: N/A*

Sulfuric Acid - YES RQ: 1000 pounds
Antimony – YES RQ: 5000 pounds
Arsenic – YES RQ: 1 pound

*Reporting not required when diameter of the pieces of solid metal released is equal to or exceeds 100 micrometers.

EPCRA Section 302 Extremely Hazardous Substance:

Sulfuric Acid – YES

EPCRA Section 313 Toxic Release Inventory:

Lead – CAS NO: 7439-92-1
Sulfuric Acid– CAS NO: 7664-93-9
Antimony – CAS NO: 7440-36-0
Arsenic – CAS NO: 7440-38-2

SECTION 16 - OTHER INFORMATION

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